

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

**EXHIBIT A**

“Compounds of Silicon” as available at

<http://www.webelements.com/webelements/compounds/text/Si/N4Si3-12033895.html>

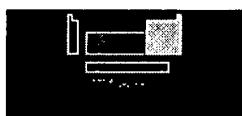
For Serial No.: 09/981,402  
Applicant(s): SATOH, Yoshihiro

**Chemistry: WebElements Periodic Table: Professional Edition: Silicon: compound data  
[silicon (IV) nitride]**

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**Silicium silicum Silicium silicio Silício silicio Kisiel**



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# Silicon

14

Si

28.0855(3)

## Compounds of silicon:

### silicon (IV) nitride

- **Formula as commonly written:**  $\text{Si}_3\text{N}_4$
- **Hill system formula:**  $\text{N}_4\text{Si}_3$
- **CAS registry number:** [12033-89-5]
- **Formula weight:** 140.283
- **Class:** nitride

## Synonyms

- silicon (IV) nitride
- silicon nitride
- trisilicon tetranitride

## Physical properties

- **Colour:** grey
- **Appearance:** crystalline solid
- **Melting point:** 1900°C
- **Boiling point:**
- **Density:** 3200 kg m<sup>-3</sup>

## Element analysis and oxidation numbers

For each compound, and where possible, a formal oxidation number for each element is given, but the usefulness of this number is limited, especially so for *p*-block elements in particular. Based upon that oxidation number, an electronic configuration is also given but note that for more exotic compounds you should view this as a guide only.

Thermodynamic properties	Element	%	Formal oxidation state	Formal electronic configuration	
<b>crystallography</b>	N	39.94	-3	[He].2s <sup>2</sup> .2p <sup>6</sup>	Cl
Crystal structure [view VR world] [view pdb image]	Si	60.06	4	[Ne]	Br

nuclear properties	<b>Synthesis</b>	Ic
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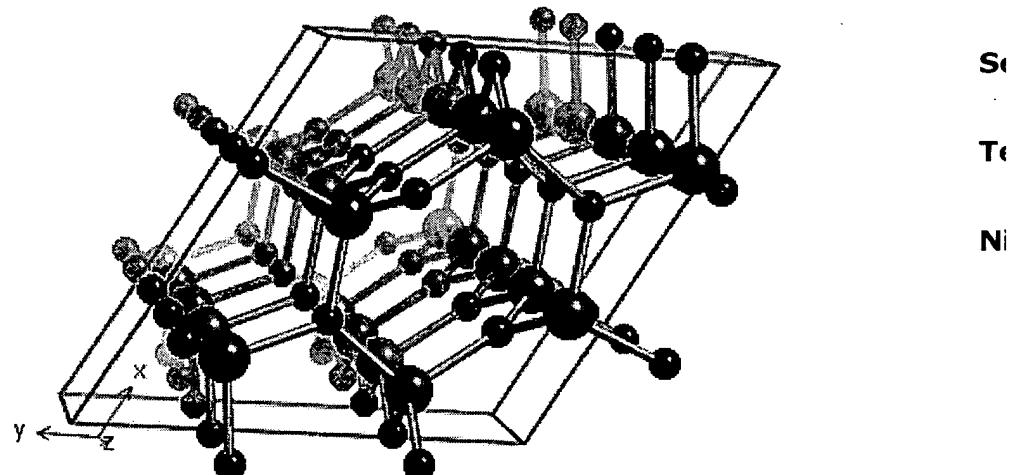
NMR	Not available	H
Naturally occurring isotopes		
Radioisotopes		

<b>WebElements</b>	<b>Solid state structure</b>	
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## Isotope pattern

What follows is the calculated isotope pattern for the  $\text{Si}_3\text{N}_4$  unit with the most intense ion set to 100%.

Formula:  $\text{Si}_3\text{N}_4$

mass	%	
140	100.0	—
141	16.7	—
142	11.1	—
143	1.2	—
144	0.4	—
145	0.0	—
146	0.0	—

## Suppliers

Coming soon....

## References

The data on these compounds pages are assembled and adapted from the primary literature and several other sources including the following.

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WebElements is the periodic table on the WWW

 WebElements™, the periodic table on the WWW, URL: <http://www.webelements.com/>  
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